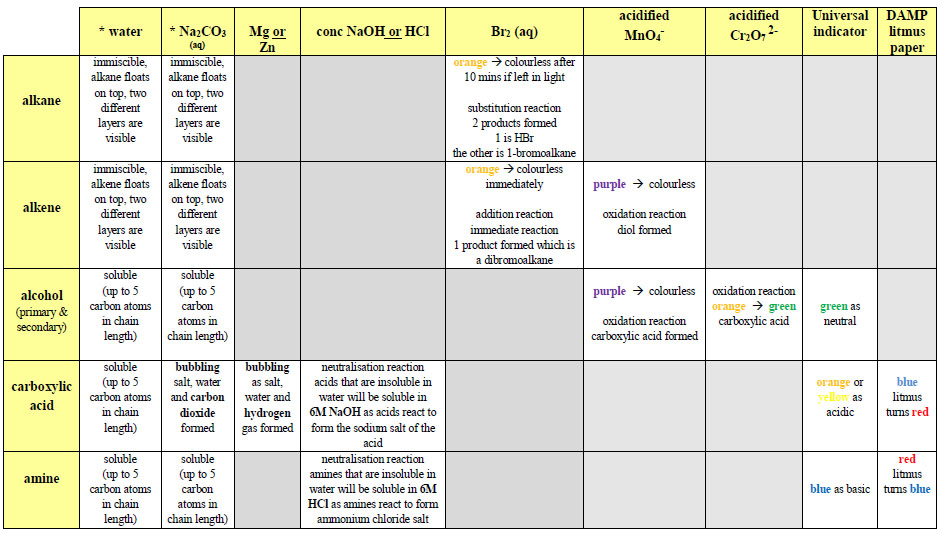
**Organic identifying (Level 2) examiners tips: Read these please!**

• this is a logical problem solving question



\* water is polar,

many organic substances eg *bromine water, hexane* are non-polar

polar and non-polar substances do not dissolve in each other, because of this two layers form as polar and non-polar layers do not mix.

**Also…”don’t be daft”**UV light has ABSOLUTELY NO effect on the reaction of acidified potassium permanganate (or dichromate) with an alkene

never ever describe a colour as clear, write COLOURLESS

for colour changes always state the initial AND final colours

litmus paper must be DAMP

be careful the CORRECT formula for potassium permanganate is KMnO4

present your answers in a WELL STRUCTURED and ORGANISED manner

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